

SAFETY DATA SHEET (SDS) **CO-METHCAL GAS STANDARDS**

Please ensure that this SDS is received by the appropriate persons

Review Date: 1	1/11/2021 v01	Emergency: 0860 02 02	2 02 Document Number: AFX-SDS-0039
1. PRODUCT	AND COMPANY DENTIFIC	ATION	- Oxygen is non-flammable, but readily
Product:	Carbon Monoxide/Methane/	'Air calibration	supports combustion. Never permit oil, grease or other readily combustible
Chemical Formula:	CO plus CH ₄ plus O ₂ plus N	2	substance to come into contact with high concentrations of oxygen. - Nitrogen does not support life. It can act
Trade Name:	105-120 ppm CO, 1.3-1.5% COMETHCAL 130/15:	CH ₄	as a simple asphyxiant by diluting the concentration of oxygen in air
	120-140 ppm CO, 1.4-1.6% COMETHCAL 150/14:	CH4	- All cylinders are transportable gas containers.
	149-180 ppm CO, 1.30-1,48 COMETHCAL 200/14: 160-240 ppm CO, 1.30-1,49	9% CH4 Adve 9% CH4 Healt 9% CH4 Finite	rseCarbon monoxide:hConcentrations in excess of 50 ppm
	COMETHCAL 400/24: 400-480 ppm CO, 2.30-2.45	5% CH ₄	ts: carbon monoxide will produce symptoms of poisoning if breathed for a sufficiently long time.
	COMETHCAL 450/14: 360-540 ppm CO, 1.30-1.49 COMETHCAL 400/24/17	9% CH₄	Methane: It is a physiologically inert, except when it
	400-480 ppm CO, 2.30-2.45% COMETHCAL 400/14/17:	CH4, 16-18% O ₂	lowers the partial pressure of oxygen in the air enough to cause systemic effects due to oxygen-deficiency
Colour Coding:	Silver body with a red shoul circular band just below the	der, and yellow red shoulder.	Oxygen: Central nervous system toxicity including
Draduat Cadar	The relevant "COMETHCAL affixed centrally to the body	" decal shall be of the cylinder.	dizziness, convulsions and loss of consciousness can occur after only 2-3 hours of exposure to pure oxygen at 2 or
Product Code:	519130-NE-A 519131-NE-A 519135-NE-A		more atmospheres. Retrosternal soreness, associated with coughing and breathing
	519136-NE-A 519134-NE-A		difficulties, made worse by smoking and exposure to cold air can occur after breathing pure oxygen at atmospheric
	519129-NE-A 519138-NE-A 519133-NE-A		pressure for several hours. <u>Nitrogen</u> :
Company Identification:	Grayston Office Park Buildir 128 Peter Road Sandown, S	ng 7 Sandton, 2196	Excessive inhalation at high concentrations can result in dizziness, nausea, vomiting, loss of consciousness
	Tel. No: (011) 490-0400 Fax No: (011) 490-0530	Chem	and death. Carbon monoxide:
Emergency	www.afrox.com	frox.linde.com Hazar	rds: No known hazards. <u>Methane</u> :
Numbers:	0860 02 02 02 (Afrox)		when methane burns in air. Oxygen:
2. HAZARD ID	ENTIFICATION		Is non-flammable, but strongly supports combustion.
Classification:	Classification under the Globa Harmonized System of classif	lly ication and	<u>Nitrogen</u> : Is relatively inert to most materials under ordinary conditions. It becomes more

	Trainionized System of Classification and
	labelling of chemicals (GHS).
Emergency	Colour: None
Overview:	Odour: None
	Taste: None
	Physical State: Compressed Gas
	Form: Gas under pressure
Main Hazards:	All cylinders are portable gas containers
	and must always be regarded as pressure
	vessels.
	- The carbon monoxide component of the
	above gas standards is a chemical
	asphyxiant.
	- Although the methane component of
	these gas mixtures will burn when ignited
	by a flame, the methane will not add
	significantly to the fire.
	o i

reactive at elevated temperatures, and combines with hydrogen, oxygen and

- Carbon monoxide combines with the

- Methane in high concentrations may cause asphyxiation. Symptoms may include loss of mobility/consciousness. - Oxygen is a local irritant to mucous membranes and, with extended continued

haemoglobin in the blood to form carboxyhaemoglobin, which is unable to transport oxygen. The symptoms of carbon monoxide poisoning are largely due to

some metals.

None.

anoxia.

Biological

Inhalation:

Hazards: Vapour



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	exposure can be destructive to lung	P201+P202: Do not handle until all safety
	tissue	precautions have been read and
	- Nitrogen acts as a simple asphyxiant	understood.
	death may result from errors in judgement.	P210: Keep away from heat/sparks/open
	confusion, or loss of consciousness which	flames/hot surfaces. No smoking.
	prevents self-rescue. At low oxygen	P260+P261: Avoid breathing
	concentrations, unconsciousness and	dust/fume/gas/mist/
	death may occur in seconds without	vapours/sprav.
	warning.	P264: If on skin (or hair) remove
Eve Contact:	No known effect.	immediately all contaminated clothing.
Skin Contact:	No known effect.	Rinse skin with water/shower.
Ingestion:	(See "Vapour Inhalation" above).	P270: Do not eat, drink or smoke when
GHS	Carbon monoxide:	using this product.
Classification:	- Flammable gas (Category 1).	P271: Use only outdoors or in a well-
	- Acute toxicity, Inhalation gas (Category	ventilated area.
	3).	P280: Wear protective gloves/protective
	- Toxic to reproduction (Category 1).	clothing/eye protection/face protection.
	- Specific Target Organ Toxicity, Repeated	Response:
	Exposure (Category 1).	P304+P340: IF INHALED: Remove
	Methane:	person to fresh air and keep comfortable
	- Flammable gas (Category 1).	for breathing.
	Oxygen:	P308 + P313: IF exposed or concerned:
	- Oxidizing gas (Category 1).	Get medical advice/attention.
	Nitrogen:	P311: Call manufacturer/supplier or the
	Non-flammable gas (Category 3)	competent authority to specify the
		appropriate source of emergency medical
GHS	\wedge \wedge \wedge	advice.
Pictogram:		P314: Get medical advice/attention if you
		feel unwell.
		P321: Specific treatment (See "First Aid
CUS Signal		Measures" below).
Worder	Daliger	P377: Leaking gas fire. Do not extinguish
GHS Hazard	Carbon monovide:	unless leak can be stopped safely.
Statements:		P381: Eliminate all ignition sources if safe
Statements.	- Extremely flammable gas	to do so.
	H280	Storage:
	- Contains das under pressure: may	P410+P403: Protect from sunlight. Store
	explode if heated	in a well- ventilated place.
	H331	P403+P233: Store in a well-ventilated
	- Toxic if inhaled	place. Keep container tightly closed.
	H360	P405: Store locked up.
	- May damage the unborn child.	Disposal:
	H372	P501: Dispose of content/container in
	- Causes damage to organs through	
	prolonged or repeated exposure.	
	Methane:	Nethone:
	H220	Brovention:
	- Extremely flammable gas.	P210: Koon away from heat/sparks/open
	H280	flames/bot surfaces. No smoking
	- Contains gas under pressure; may	Bosponso:
	explode if heated.	P377: Leaking gas fire Do not extinguish
	Oxygen:	unless leak can be stonned safely
	H270	P381: Eliminate all ignition sources if safe
	 May cause or intensify fire; oxidizer. 	to do so
	H280	Storage:
	- Contains gas under pressure; may	P410+P403: Protect from sunlight Store
	explode if heated.	in a well- ventilated place
	Nitrogen:	Oxygen:
	H280	Prevention:
	 Contains gas under pressure; may 	P220: Keep/Store away from combustible
	explode if heated.	materials. Manufacturer/supplier or the
GHS	Carbon monoxide:	competent authority to specify other
Precautionary	Prevention:	incompatible materials.
Statements:		
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	P244: Keep valves and fittings free froil and grease. Response: P307+P376: In case of fire: Stop leads a safe to do so.	rom Skin Contact: k if	Adverse effects not expected from this product. <u>Carbon monoxide</u> : Not relevant, due to the form of the product.
Other Hazards that do not result in classification:	Storage: P410+P403: Protect from sunlight. S in a well- ventilated place. <u>Nitrogen:</u> P410+P403: Protect from sunlight. S in a well- ventilated place. <u>Carbon monoxide:</u> - Chemical asphyxiant. - Exposure to low concentrations for extended periods may result in dizziness or unconsciousness and ma lead to death.	Store Store Ingestion: ay	Methane:Adverse effects not expected from this product.Oxygen:Adverse effects not expected from this product.Nitrogen:Adverse effects not expected from this product.Carbon monoxide:Ingestion is not considered a potential route of exposure.Methane:Ingestion is not considered a potential
3. COMPOSIT	ION OF INGREDIENTS		route of exposure.
Chemical name	: Carbon monoxide (CO) Methane (CH ₄) Oxygen (O ₂) Nitrogen (N ₂)		<u>Oxygen</u> . Ingestion is not considered a potential route of exposure. <u>Nitrogen</u> : Ingestion is not considered a potential
e,	Methane Oxygen	Inhalation:	<u>Carbon monoxide</u> : - Remove victim to uncontaminated area
CAS No:	<u>Carbon monoxide</u> : 630-08-0 <u>Methane</u> : 74-82-8 Oxvgen: 7782-44-7		wearing self-contained breatning apparatus. - Keep victim warm and rested.
UN No:	<u>Nitrogen</u> : 7727-37-9 <u>Carbon monoxide</u> : 1016 <u>Methane</u> : 1971 Oxygen: 1072		- Apply artificial respiration if breathing stopped. <u>Methane</u> :
ERG No:	<u>Nitrogen</u> : 1066 <u>Carbon monoxide</u> : 119 <u>Methane</u> : 115 <u>Oxygen</u> : 122		asphyxiation. - Symptoms may include loss of mobility/consciousness. Victim may not be aware of asphyxiation. Remove victim
Hazard class:	<u>Nitrogen</u> : 121 <u>Carbon monoxide</u> : 2.2 <u>Methane</u> : 2.1 <u>Oxygen</u> : 2.2		to uncontaminated area wearing self- contained breathing apparatus. - Keep victim warm and rested. - Call a doctor.
Hazchem Warni	ng: <u>Carbon monoxide</u> : Toxic gas <u>Methane</u> : Flammable gas <u>Oxygen</u> : Non-flammable gas <u>Nitrogen</u> : Non-flammable gas		 Apply artificial respiration if breathing stopped. <u>Oxygen</u>: Move the exposed person to fresh air at once.
			Nitrogen: - In high concentrations may cause
4. FIRST AID	MEASURES		asphyxiation
Eye contact:	Carbon monoxide: Adverse effects not expected from this product. <u>Methane:</u> Adverse effects not expected from this product. <u>Oxygen:</u> Adverse effects not expected from this product. Nitrogen:	s s	 Symptoms may include loss of mobility/consciousness. Victim may not be aware of asphyxiation. Remove victim to uncontaminated area wearing self-contained breathing apparatus. Keep victim warm and rested. Call a doctor. Apply artificial respiration if breathing stopped.



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5. FIRE-FIGHTING MEASURES

Suitable	Carbon monoxide+Methane: Dry
extinguishing	chemical, Water spray or fog, Foam.
media:	<u>Oxygen</u> : Water.
	Nitrogen: Material will not burn. In case of
	fire in the surroundings: use appropriate
	extinguishing agent.
Unsuitable	Carbon monoxide+Methane: Carbon
extinguishing	dioxide.
media:	<u>Oxygen</u> : None.
	<u>Nitrogen</u> : None.
Specific	Carbon monoxide: None
Hazards:	Methane: Incomplete combustion may
	form carbon monoxide.
	Oxygen: Supports combustion.
	<u>Nitrogen</u> : None.
Special fire	 In case of fire: Stop leak if safe to do so.
fighting	- Do not extinguish flames at leak because
procedures:	possibility of uncontrolled explosive re-
	ignition exists. Continue water spray from
	protected position until container stays
	cool. Use extinguishants to contain the fire.
	Isolate the source of the fire or let it burn
	out.
Special	Firefighters must use standard protective
protective	equipment including flame retardant coat,
equipment	helmet with face shield, gloves, rubber
for	boots, and in enclosed spaces, self-
firefighters:	contained breathing apparatus.

6. ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures:	 Evacuate area. Provide adequate ventilation. Consider the risk of potentially explosive atmospheres. In case of leakage, eliminate all ignition sources. Monitor the concentration of the released product. Prevent from entering sewers, basements and workpits, or any place where its accumulation can be dangerous. Wear self-contained breathing apparatus when entering area unless atmosphere is proved to be safe.
Environmental Precautions:	 Respiratory protective devices - Self- contained open circuit compressed air breathing apparatus with full face mask - Requirements, testing, marking. Prevent further leakage or spillage if safe to do so. Prevent waste from contaminating the surrounding environment. Prevent soil and water pollution. Dispose of contents/container in accordance with local/regional/national/international regulations. Contact supplier for any special requirements.

Methods and Provide adequate ventilation. Eliminate material for sources of ignition. containment and cleaning

7. HANDLING AND STORAGE

	<u> </u>
Safe Handling:	- Only experienced and properly instructed persons should handle gases under
	Keen every from oil groops heat het
	- Keep away from oil, grease neat, not
	surfaces, sparks, open flames and other
	ignition sources.
	- No smoking.
	 Never use direct flame or electrical
	heating devices to raise the pressure of a
	container.
	- Use only non-sparking tools. Use only
	explosion-proof equipment.
	- Use only with equipment cleaned for
	oxygen service and rated for the pressure.
	- Wear leather safety gloves and safety
	shoes when handling cylinders.
	- Protect cylinders from physical damage:
	do not drag, roll, slide or drop. While
	moving cylinder, always keep in place
	removable valve cover. Never attempt to lift
	a cylinder by its cap: the cap is intended
	solely to protect the valve. When moving
	cylinders, even for short distances, use a
	cart (trolley hand truck etc.) designed to
	transport cylinders
	- Never insert an object (e.g. wrench
	screwdriver, pry bar) into can openings:
	doing so may damage the valve and cause
	a leak
	- Use an adjustable strap wrench to
	remove over-tight or rusted caps. Slowly
	open the value. If the value is hard to open
	discontinue use and contact your supplier
	- Close the container valve after each use:
	keen closed even when empty
	- Secure cylinders in an unright position at
	all times, close all valves when not in use
	l covo volvo protoction cons in placo until
	the container has been secured against
	aither a wall or banch or placed in a
	container stand and is ready for use
	Nover apply flame or localized heat
	directly to any part of the container. High
	temperatures may demage the container. Fight
	emperatures may damage the container
	to foil promoturoly, vonting the container
	to fail prematurely, venting the container
	Contents.
	- Use only oxygen approved lubricants
	anu sealants.
	- Neep container valve outlets clean and
	mee norm contaminates particularly oil and

water. - Do not remove or deface labels provided by the supplier for the identification of the



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Conditions for safe storage, including any incompatibilit ies:	 container contents. Damaged valves should be reported immediately to the supplier. Never attempt to repair or modify container valves or safety relief devices. For other precautions in using this product, see "Other Information" below. All electrical equipment in the storage areas should be compatible with the risk of a potentially explosive atmosphere. Segregate from flammable gases, oxidant gases and other oxidants being stored. Containers should not be stored in conditions likely to encourage corrosion. Stored containers should be periodically checked for general conditions and leakage. Container valve guards or caps should be in place. Store containers in location free from fire risk and away from sources of heat and ignition. Avoid asphalted locations for storage, transfer and use (implice risk is prime risk is place).
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transfer and use (ignition risk if spilt). - Keep away from combustible material.

8. EXPOSURE CONTROLS AND PERSONAL PROTECTION

Occupational	Prolonged exposure to low concentrations
Exposure	of carbon monoxide may cause permanent
Hazards	harmful effects.
(HCS):	
Engineering	- Consider a work permit system e.g., for
Control	maintenance activities.
Measures:	- Ensure adequate air ventilation. Provide
	adequate general and local exhaust
	ventilation.
	 Keep concentrations well below lower
	explosion limits. Gas detectors should be
	used when quantities of oxidizing gases,
	flammable gases or vapours may be
	released.
	- Provide adequate ventilation, including
	appropriate local extraction, to ensure that
	the defined occupational exposure limit is
	not exceeded.
	- Systems under pressure should be
	regularly checked for leakages.
	- Product to be handled in a closed system.
	Use only permanent leak tight installations
	(e.g., welded pipes).
	- Take precautionary measures against
	static discharges.
Personal	- A risk assessment should be conducted
protection:	and documented in each work area to
-	assess the risks related to the use of the
	product and to select the PPE that
	matches the relevant risk.
	- Keep self-contained breathing apparatus
	readily available for emergency use.
	- Keep suitable chemically resistant
	protective clothing readily available for

	emergency use.
	 Personal protective equipment for the
	body should be selected based on the task
	being performed and the risks involved
	being performed and the fisks involved.
	Protect eyes, face and skin from contact
	with product. Refer to local regulations for
	restriction of emissions to the atmosphere.
	- See section 13 for specific methods for
	weste ges treatment
_	waste gas treatment.
Eyes:	Wear safety glasses when handling
	cylinders; vapor-proof goggles and a face
	shield during cylinder changeout or
	whenever contact with product is possible.
Hands.	Wear working gloves when handling gas
nanao.	containara
Dealer	Containers.
воау	wear metatarsal shoes and work gloves for
protection:	cylinder handling, and protective clothing
	where needed. Wear appropriate chemical
	aloves during cylinder changeout or
	wherever contact with
	product is possible
F	
Feet:	Wear safety shoes while handling
	containers.

9. PHYSICAL AND CHEMICAL PROPERTIES

Chemical Name:	Carbon Monoxide
Chemical Symbol:	СО
Appearance Physical	Gas
state: Form: Colour:	Compressed gas Colourless
Odour Odour Threshold:	Odour threshold is subjective and is inadequate to warn of over
pH: Melting Point: Boiling Point:	Not applicable -205 °C -191.5 °C
Sublimation Point:	Not applicable
Critical Temp. (°C):	-140.0 °C
Flash Point: Evaporation	Not applicable.
Rate:	Not applicable.
(solid, gas):	Flammable gas.
limit - upper (%):	74% (V)
Flammability limit – lower (%):	10.9% (V)
Vapour pressure:	> 101.325 kPa (20 °C)
Vapour density (air=1):	0.968 Air=1
Relative density:	0.97 (20 °C)



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Solubility in Water: Partition	29 g/l (20 °C)	Decompositio Temperature: Viscosity	on Not applicable.
coefficient (n- octanol/water):	1.78	Kinematic viscosity:	Not applicable.
Autoignition Temperature:	605 °C	Dynamic viscosity:	Not applicable.
Decomposition Temperature:	400 °C	Explosive properties:	Not applicable.
Viscosity Kinematic viscosity:	Not applicable.	Properties:	None.
Dynamic viscosity:	Not applicable.	Chemical Name:	<u>Oxygen</u>
Explosive properties:	Not applicable.	Chemical Symbol:	O ₂
Oxidising Properties:	None.	Appearance Physical state:	Gas
Chemical Name:	Methane	Form: Colour:	Compressed gas Colourless
Chemical Symbol:	CH ₄	Odour: Odour	Odourless Odour threshold is subjective and is
Appearance Physical	0	Threshold: pH:	inadequate to warn of over exposure. Not applicable.
state: Form:	Gas Compressed gas	Melting Point Boiling Point	: −218.4 °C −183 °C
Colour: Odour:	Colourless Odourless	Sublimation Point:	Not applicable.
Odour Threshold:	Odour threshold is subjective inadequate to warn of over	e and is Critical Temp (°C):	-118.0 °C
pH:	exposure. Not applicable.	Flash Point: Evaporation	Not applicable.
Melting Point: Boiling Point:	-182.47 °C -161.48 °C	Rate: Flammability	This product is not flammable
Sublimation Point:	Not applicable.	(solid, gas): Flammability	
Critical Temp. (°C):	-82.0 °C	limit - upper (%):	Not applicable.
Evaporation	Not applicable.	Flammability limit – lower (%)-	Not applicable.
Flammability (solid, gas):	Flammable gas.	Vapour pressure:	4,053 kPa (-124.1 °C)
Flammability limit - upper	17 %(V)	Vapour densi (air=1):	ty 1.1 (0 °C) Air=1
(%): Flammability limit – lower	4 4 %(\/)	Relative density: Solubility in	1.1 (0 °C, Reference material: Water)
(%): Vapour		Water: Partition	39 mg/l
pressure: Vapour density	Not applicable.	coefficient (n octanol/water	- Not known.):
(air=1): Relative	0.42 (25 °C)	Autoignition Temperature	Not applicable.
density: Solubility in Water:	22 mg/l (25 °C)	Decompositio Temperature: Viscosity	on Not known.
Partition coefficient (n-	1.09	Kinematic viscosity:	Not applicable.
octanol/water): Autoignition	537 °C	Dynamic viscosity:	Not applicable.
remperature:		Explosive properties:	Not applicable.



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Oxidising Properties:	Oxidising.	Chemical stability:	Stable under normal conditions.
Chemical	<u>Nitrogen</u>	Possibility of hazardous	Carbon monoxide: - Can form a potentially explosive atmosphere in air. May react violently with
Chemical Symbol:	N ₂	reactions.	oxidants. Methane:
Appearance Physical	Con		- Stable under normal conditions. Oxygen:
state: Form:	Compressed gas		 Violently oxidises organic material. May react violently with combustible materials
Colour: Odour:	Colourless Odourless		May react violently with reducing agents. <u>Nitrogen</u> :
Odour Threshold:	Odour threshold is subjective and is inadequate to warn of over exposure.	Conditions to	None. - Avoid moisture in the installation.
pH: Melting Point: Boiling Point:	Not applicable. -210.01 °C -196 °C	avoid:	- Keep away from neat, not surfaces, sparks, open flames and other ignition sources
Sublimation Point: Critical Temp	Not applicable.	Incompatible Materials:	- No smoking. - Air and oxidisers. - Moisture
(°C): Flash Point:	-147.0 °C Not applicable.	materialo.	- Combustible materials. - Reducing Agents.
Evaporation Rate:	Not applicable.		- Keep equipment free from oil and grease.
Flammability (solid, gas): Flammability	This product is not flammable.	Hazardous Decomposition of Products:	Under normal conditions of storage and use, hazardous decomposition products should not be produced
limit - upper (%):	Not applicable.		
Flammability limit – lower	Not applicable.	11. TOXOLOG Acute Toxicity:	ICAL INFORATION Not classified
Vapour pressure:	Not applicable.	Skin & eye cont Chronic Toxicit	tact: Not classified y: Not classified Vet classified Not classified
Vapour density (air=1): Relative	0.97	Mutagenicity: Reproductive H	Not classified Not classified azards: Not classified
density: Solubility in	0.8		
Water: Partition	20 mg/i	12. ECOLOGIC	CAL INFORMATION
coefficient (n- octanol/water):	0.67	Persistence	product. Not readily biodegradable. Not applicable
Temperature:	Not applicable.	and degradability:	for inorganic gases.
Temperature: Viscosity	Not known.	Mobility in soil: Ecology	No data available.
Kinematic viscosity:	Not applicable.	soil:	is unlikely to cause ground or water pollution.
Dynamic viscosity: Explosive	0.171 mPa.s (10.9 °C)	Results of PBT and	Not classified as PBT or vPvB.
properties: Oxidisina	Not applicable.	vPvB assessment:	
Properties:	Not applicable.	Other adverse effects:	No ecological damage caused by this product.
		Effect on	None.

ozone layer:

Effect on the

global

warming:

Carbon monoxide: 1.9 [CO2=1]

Methane:

10. STABILITY AND REACTIVITY

Reactivity:

No reactivity hazard other than the effects described in sub-section below.



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	Global warming potential: 25 [Contains greenhouse gas(es) discharged in large quantities contribute to the greenhouse e Oxygen:	CO2=1] ERG No.: When may effect. Class:	<u>Carbon monoxide</u> : 119 <u>Methane</u> : 115 <u>Oxygen</u> : 122 <u>Nitrogen</u> : 121 Carbon monoxide: 2.1
	No known effects from this pro <u>Nitrogen</u> : None.	oduct.	<u>Methane</u> : 2.1 <u>Oxygen</u> : 2.2 <u>Nitrogen</u> : 2.2
13. DISPOSAL CONSIDERATIONS		Subsidiary	Risk: <u>Carbon monoxide</u> : Toxic gas <u>Methane</u> : Flammable gas
Disposal Methods:	- Dispose of container via sup - For more detailed informatio guidance contact the nearest branch.	plier only. n or Label: Afrox	<u>Oxygen</u> : Non-flammable gas <u>Nitrogen</u> : Non-flammable gas <u>Carbon monoxide</u> : Toxic gas <u>Methane</u> : Flammable gas <u>Oxygen</u> : Non-flammable gas
Disposal of Packaging:	supplier and the disposal of th	e Air Transc	Nitrogen: Non-flammable gas
	containers must only be hand	led by the	Code: Carbon monoxide: 1016
14 TRANSPOL			<u>Methane</u> : 1971 <u>Oxygen</u> : 1072 <u>Nitrogen</u> : 1066
14. TRANSPORT INFORMATION Road Transportation		Class:	Carbon monoxide: 2.1
UN No.:	Carbon monoxide: 1016 Methane: 1971		<u>Oxygen</u> : 2.2 <u>Nitrogen</u> : 2.2
Shipping Name	<u>Oxygen</u> : 1072 <u>Nitrogen</u> : 1066 : <u>Carbon monoxide</u> : Com gas	pressed	Nethane: Flammable gas <u>Oxygen</u> : Non-flammable gas <u>Nitrogen</u> : Non-flammable gas
FRG No -	<u>Methane</u> : Compressed <u>Oxygen</u> : Compressed g <u>Nitrogen</u> : Compressed g Carbon monovide: 119	gas Packaging as instructions gas	<u>Carbon monoxide</u> : s: - Cargo: Forbidden - Passenger: Forbidden Methane:
	Methane: 115 Oxygen: 122 Nitrogen: 121		- Cargo: Forbidden - Cargo aircraft only: Allowed (200) - Passenger: Forbidden
Class:	Carbon monoxide: 2.3 <u>Methane</u> : 2.1 <u>Oxygen</u> : 2.2 Nitrogen: 2.2		<u>Oxygen</u> : - Cargo: Allowed (200) - Cargo aircraft only: Allowed - Passenger: Allowed (200)
Subsidiary Risk	Carbon monoxide: Pois <u>Methane</u> : Flammable ga <u>Oxygen</u> : Non-flammable	on gas as e gas	<u>Nitrogen</u> : - Cargo: Allowed (200) - Passenger: Allowed (200)
Hazchem Warni	<u>Nitrogen</u> : Non-flammab ing: <u>Carbon monoxide</u> : - Flammable gas - Poison gas <u>Methane</u> : - Flammable gas <u>Oxygen</u> : - Oxidizing gas <u>Nitrogen</u> :	e gas Maximum o allowed:	- Cargo: Forbidden - Passenger: Forbidden <u>Methane:</u> - Cargo: 150 kg - Passenger: Forbidden <u>Oxygen</u> : - Cargo: 100 kg - Passenger: 75 kg
Soo Transport	- Non-flammable gas		<u>Nitrogen:</u> - Cargo: 150 kg
	Carbon monovide: 1016		- Passenger: 75 kg
IMDG.	<u>Methane</u> : 1971 <u>Oxygen</u> : 1072		
Shinning Nama	Nitrogen: 1066	15. KEGUL	ATORY INFORMATION
	gas <u>Methane</u> : Compressed <u>Oxygen</u> : Compressed g <u>Nitrogen</u> : Compressed g	gas SANS 1023 as SANS 1023 as SANS 1023 Jas ISO 11014	4 – Supplement
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Please ensure that this SDS is received by the appropriate persons

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16. OTHER INFORMATION

- Ensure all national/local regulations are observed.
- Ensure users and relevant persons understand the asphyxiation hazard.
- Regularly check suppliers information sources for updated versions of SDS's.

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BIBLIOGRAPHY

Compressed Gas Association, Arlington, Virginia Handbook of Compressed Gases - 3rd Edition Matheson. Matheson Gas Data Book - 6th Edition SABS 0625 - Labelling of Dangerous Substances

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