

# Please ensure that this SDS is received by the appropriate persons

#### Review Date: 29/10/2021 v01

Emergency: 0860 02 02 02

Vapour Inhalation: Document Number: AFX-SDS-0042

constituent of the acid-base mechanism that controls the pH of the blood. - At concentrations of 10 % and above of

carbon dioxide, unconsciousness can

#### 1. PRODUCT AND COMPANY IDENTIFICATION

Product: Chemical Formula:	Foodfresh Packaging Gas Mixtures CO <sub>2</sub>
Trade Name: Colour Coding:	FoodFresh 1 Light Brunswick Green (5540-G30Y) body with the relevant grades stenciled on their bodies and Cloud white (0505- G20Y) valve guard
Product Code: Company Identification:	518701-SE-C African Oxygen Limited Grayston Office Park Building 7 128 Peter Road Sandown, Sandton, 2196 Tel. No: (011) 490-0400 Fax No: (011) 490-0530 Email: <u>customer.service@afrox.linde.com</u> www.afrox.com
Emergency Numbers:	0860 02 02 02 (Afrox)

### 2. HAZARD IDENTIFICATION

Classification:	Classification under the Globally
	Harmonized System of classification and
	labelling of chemicals (GHS).
Emergency	Colour: None
Overview:	Odour: None
	Taste: Acidic
	Physical State: Compressed Gas
	Form: Gas under pressure
Main Hazards:	- The listed grade of Foodfresh do not
	support life.
	- It can act as simple asphyxiants by
	diluting the concentration of oxygen in air
	below the levels necessary to support life.
	- As it is heavier than air it will
	concentrate at lower levels.
Adverse Health	- The carbon dioxide component
Effects:	contained in the relevant grade of
	Foodfresh acts as a stimulant and a
	depressant on the central nervous
	system.
	- Increases in heart rate and blood
	pressure have been noted at a
	concentration of 7.6 percent, and
	dyspnea (laboured breathing), headache,
	dizziness and sweating occur if exposure
	at that level is prolonged.
Chemical	<ul> <li>Carbon dioxide components for</li> </ul>
Hazards:	Foodfresh 1 are relatively non-reactive
	and non-toxic.
	<ul> <li>They will not burn or support</li> </ul>
	combustion.
Biological	<ul> <li>The greatest physiological effect of</li> </ul>
Hazards:	carbon dioxide is to stimulate the
	respiratory centre, thereby controlling the
	volume and rate of respiration.
	<ul> <li>It is able to cause dilation and</li> </ul>
	constriction of blood vessels and is a vital

	<ul> <li>Impairment in performance has been noted during prolonged exposure to concentrations of 3% carbon dioxide even when the oxygen concentration was 21 %.</li> </ul>
Eye Contact:	No known effect.
Skin Contact:	No known effect.
Ingestion:	(See "Vapour Inhalation" above).
GID Classification:	Respiratory sensilizers (Category 1).
GHS	<b>^</b>
Pictogram:	
GHS Signal	Danger
Words:	H280
Statements:	- Contains das under pressure may
	explode if heated.
GHS	Storage:
Precautionary	P410+P403: Protect from sunlight. Store
Statements:	in a well-ventilated place.
	Prevention: P282: Wear cold insulating gloves/face
	shield/eve protection.
	Response:
	- P336: Thaw frosted parts with
	lukewarm water. Do not rub affected
	area.
	- roio. Get immediate medical
	Storage:
	P403: Store in well-ventilated place.
Other Hazards	- Asphyxiant in high concentrations.
that do not	- Contact with liquid may cause cold
result in	burns/frostbite.
classification:	

# 3. COMPOSITION OF INGREDIENTS

Chemical name:	Carbon dioxide (CO <sub>2</sub> )
Chemical family:	Carbon dioxide
CAS No:	124-38-9
UN No:	1013
ERG No:	120
Hazard class:	2.2
Hazchem Warning:	Non-flammable gas

## 4. FIRST AID MEASURES

Eye contact: - Rinse the eye with water immediately. - Remove contact lenses, if present and easy to do. Continue rinsing. Flush thoroughly with water for at least 15 minutes.



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Skin Contact: Ingestion: Inhalation:	<ul> <li>Get immediate medical assistance medical assistance is not immediavailable, flush an additional 15 minute Contact with evaporating liquid may controstibite or freezing of skin.</li> <li>Ingestion is not considered a potential of exposure.</li> <li>In high concentrations may consphyxiation.</li> <li>Symptoms may include loss mobility/consciousness.</li> <li>Victim may not be aware of asphyxiation.</li> </ul>	e. If Environmental ately Precautions: s. ause oute ause Methods and material for of containment and cleaning tion. up:	<ul> <li>which is very cold and can cause frostbite?</li> <li>Evacuate the area.</li> <li>Prevent further leakage by shutting off the source of the spill if safe to do so.</li> <li>Restrict access to the area until completion of the clean-up procedure.</li> <li>Ventilate the area using forced draft if necessary.</li> <li>Shut off the source of the escaping nitrogen.</li> <li>Provide adequate ventilation.</li> </ul>
	wearing self-contained breathing appar	atus.	
	<ul> <li>Keep victim warm and rested.</li> <li>Call a doctor.</li> </ul>	7. HANDLING	G AND STORAGE
	<ul> <li>Apply artificial respiration if brea stopped.</li> <li>Low concentrations of CO<sub>2</sub> c increased respiration and headache.</li> </ul>	hing Safe Handling: ause	<ul> <li>Avoid breathing gas.</li> <li>Do not get in eyes, on skin, or on clothing</li> <li>This gas is heavier than air and in an enclosed space tends to accumulate near the floor, displacing air and pushing it</li> </ul>
	TING	_	atmosphere near the floor. Ventilate space
5. FIRE-FIGH Suitable extinguishing media: Unsuitable extinguishing media: Specific Hazards: Special fire fighting procedures: Special protective equipment for firefighters:	<ul> <li>TING</li> <li>Material will not burn. In case of fire in surroundings: use appropriate extinguishing agent. None.</li> <li>The range of Foodfresh mixtures do a support life.</li> <li>They can act as simple asphyxiants to diluting the concentration of oxygen in air below the levels to support life.</li> <li>In case of fire: Stop leak if safe to do</li> <li>Continue water spray from protected position until container stays cool.</li> <li>Use extinguishants to contain the fire</li> <li>Isolate the source of the fire or let it bout.</li> <li>Firefighters must use standard protectied equipment including flame retardant con helmet with face shield, gloves, rubber boots, and in enclosed spaces, self-contained breathing apparatus.</li> </ul>	the not y the so. urn ve at,	<ul> <li>almosphere hear the noor. Vertifate space before entry. Verify sufficient oxygen concentration.</li> <li>Wear leather safety gloves and safety shoes when handling cylinders.</li> <li>Protect cylinders from physical damage; do not drag, roll, slide or drop. While moving cylinder, always keep in place removable valve cover.</li> <li>Never attempt to lift a cylinder by its cap; the cap is intended solely to protect the valve.</li> <li>When moving cylinders, even for short distances, use a cart (trolley, hand truck, etc.) designed to transport cylinders.</li> <li>Never insert an object (e.g., wrench, screwdriver) into cap openings; doing so may damage the valve and cause a leak.</li> <li>Use an adjustable strap wrench to remove over-tight or rusted caps.</li> <li>Slowly open the valve. If the valve is hard to open, discontinue use and contact your supplier.</li> <li>Close the container valve after each use; keep closed even when empty.</li> <li>Never apply flame or localized heat directly to any part of the container. High</li> </ul>
o. ACCIDENT Personal precautions, protective equipment and emergency procedures:	AL RELEASE - Evacuate area Provide adequate ventilation Prevent from entering sewers and workpits, or any place where its accumulation can be dangerous Wear self-contained breathing appar when entering area unless atmospher is proved to be safe Rapid release of gaseous carbon dio through a pressure relief device or val can result in the formation of dry ice.	Atus for safe storage, including any incompatibilit kide ies:	<ul> <li>temperatures may damage the container high temperatures may damage the container and could cause the pressure relief device to fail prematurely, venting the container contents.</li> <li>Containers should not be stored in conditions likely to encourage corrosion.</li> <li>Stored containers should be periodically checked for general conditions and leakage.</li> <li>Container valve guards or caps should be in place.</li> </ul>



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	<ul> <li>Store containers in location free from risk and away from sources of heat and ignition.</li> </ul>	fire Odour Thresho	ld:	Odour threshold is subjective and is inadequate to warn of over
	- Keep away from combustible materia	pH: Melting Point: Boiling Point:		exposure. 3.2 - 3.7 -56.6 °C -78 5 °C
8. EXPOSUR	RE CONTROLS AND PERSONAL	Sublimation Po Critical Temp. ( Flash Point:	int: °C):	-78.5 °C 31.0 °C Not applicable.
Occupational Exposure	- TWA 5000 ppm - STEL 15000 ppm	Evaporation Ra Flammability (s	te: olid, gas):	Not applicable. This product is not flammable.
Hazards (HCS): Engineering	Consider a work parmit avatam a g f	Flammability lir (%):	nit - upper	Not applicable.
Control Measures:	- Consider a work permit system e.g., i maintenance activities. - Ensure adequate air ventilation	(%):	nit – Iower	Not applicable.
incubureo.	<ul> <li>Oxygen detectors should be used wh asphyxiating gases may be released.</li> <li>Provide adequate ventilation, includin</li> </ul>	en Vapour density Relative density g Solubility in Wa	(air=1): /: /:	1.522 (21 °C) 1.512 (-56.6 °C) Completely soluble.
	appropriate local extraction, to ensure t	hat Partition coeffic	cient (n-	0.83
	not exceeded.	Autoignition Te	mperature:	Not applicable.
	<ul> <li>Systems under pressure should be regularly checked for leakages. Prefera use permanent leak tight connections ( welded pipes)</li> </ul>	bly Temperature: e.g., Viscosity		Not known.
	<ul> <li>Do not eat, drink or smoke when usin the product.</li> </ul>	g viscosity: Dynamic		No data available.
	- CO <sub>2</sub> detectors should be used when (	CO <sub>2</sub> viscosity:		0.07 mPa.s (20 °C)
Personal protection:	<ul> <li>A risk assessment should be conduct and documented in each work area to</li> </ul>	ed Oxidising Prope	erties: erties:	Not applicable
	assess the risks related to the use of the			
	the relevant risk.	10. STABILITY	AND REAC	TIVITY
	<ul> <li>Keep self-contained breathing appara readily available for emergency use.</li> <li>Personal protective equipment for the</li> </ul>	tus Reactivity: Chemical	No reactivit described i Stable unde	ty hazard other than the effects n sub-section below. er normal conditions.
Eyes:	body should be selected based on the being performed and the risks involved Wear safety glasses when handling	ask stability: Possibility of hazardous	None.	
	cylinders; vapor-proof goggles and a fa shield during cylinder changeout or whenever contact with product is possi	ce reactions: Conditions to ble. avoid:	None.	
Hands:	Wear working gloves while handling containers.	Incompatible Materials:	No reaction dry or wet o	n with any common materials in conditions.
Body protection:	No special precautions.			
Feet:	Wear safety shoes while handling containers.	Hazardous Decomposition	Under norn use, hazaro	nal conditions of storage and dous decomposition products

# 9. PHYSICAL AND CHEMICAL PROPERTIES

Chemical Name: Chemical Symbol:	Carbon Dioxide CO <sub>2</sub>
Appearance	
Physical state:	Gas
Form:	Liquefied gas
Colour:	Colourless
Odour:	Odourless

# **11. TOXOLOGICAL INFORATION**

Acute Toxicity: Skin & eye contact: Chronic Toxicity: Carcinogenicity: Mutagenicity: Reproductive Hazards:

of Products:

Not classified Not classified Not classified Not classified Not classified Not classified

should not be produced.



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#### **12. ECOLOGICAL INFORMATION**

Toxicity:	No ecological damage caused by this product.
Persistence and	No ecological damage caused by this product.
degradability:	
Mobility in soil:	No data available.
Ecology - soil:	No ecological damage caused by this product.
Results of PBT and vPvB	Not classified as PBT or vPvB.
assessment:	
Other adverse effects:	No ecological damage caused by this product.
Effect on	None.
Effect on the global warming:	When discharged in large quantities may contribute to the greenhouse effect.

#### **13. DISPOSAL CONSIDERATIONS**

Disposal	- Dispose of container via supplier only.
Methods:	<ul> <li>For more detailed information or guidance contact the nearest Afrox branch.</li> </ul>
Disposal of Packaging:	The container is the property of the supplier and the disposal of the containers must only be handled by the supplier.

# 14. TRANSPORT INFORMATION Road Transportation

UN No: Shipping Name: ERG No: Class: Subsidiary Risk: Hazchem Warning: Sea Transportation	1013 Carbon Dioxide 120 2.2 Chemical Asphyxiant Non-flammable gas
IMDG: Shipping Name: ERG No: Class: Subsidiary Risk: Label:	1013 Carbon Dioxide 120 2.2 Chemical Asphyxiant Non-flammable gas
Air Transportation	-
ICAO/IATA Code: Class: Subsidiary risk: Packaging instructions: Maximum quantity allowed:	1013 2.2 Chemical Asphyxiant - Cargo: 200 kg - Passenger: 200 kg - Cargo: 150 kg - Passenger: 75 kg

#### **15. REGULATORY INFORMATION**

SANS11014:2010 Edition 1:	Safety data sheet for chemical products - Content and order of sections.
SANS 10228:2012 Edition 6:	The identification and classification of dangerous goods for transport by
SANS 10234-2010	road and rail modes.
Edition 2:	classification and labelling of chemicals (GHS).
SUPPLEMENT TO SANS 10234 Edition 1:	List of classification and labelling of chemicals in accordance with the Globally Harmonized System (GHS).

#### **16. OTHER INFORMATION**

- Ensure all national/local regulations are observed.Ensure users and relevant persons understand the
- asphyxiation hazard.
- Regularly check suppliers information sources for updated versions of SDS's.
- Revision Date:
- 29/10/2021 v01

#### BIBLIOGRAPHY

Compressed Gas Association, Arlington, Virginia Handbook of Compressed Gases - 3rd Edition Matheson. Matheson Gas Data Book - 6th Edition SABS 0625 - Labelling of Dangerous Substances

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