

Please ensure that this SDS is received by the appropriate persons

Review Date: 30/03/2022 v01 Emergency: 0860 02 02 02 Document Number: AFX-SDS-0028

1. PRODUCT	AND COMPANY IDENTIFICATION	
Product	Handigas	
	Liquefied Petroleum Gas (LPG or LPGas)	
Chemical Formula	C ₃ H _{8 +} C ₄ H _{10 +} C ₃ H ₆	
Trade Name	Handigas LPG	
Colour Coding	Plascon Dark Admiralty Grey (SABS 1091 – G.12) body, with a Handigas decal affixed to the cylinder. All cylinders fitted with an internal eductor	
	tube for liquid withdrawal are clearly marked with two Yellow (B.49) stripes painted diametrically opposite each other along the length of the cylinder	
Product Code	3634 LC 9kg 3634 LD 14kg 3634 LE 19kg 3634 LF 48kg 3634 L F-W 48kg wet	
Company Identification	African Oxygen Limited Grayston Office Park Building 7 128 Peter Road Sandown, Sandton, 2196 Tel. No: (011) 490-0400 Fax No: (011) 490-0530 Email: customer.service@afrox.linde.com www.afrox.com	
Emergency Numbers	0860 02 02 02 (Afrox)	

2. HAZARD IDENTIFICATION		
Classification	- Classification under South African Hazardous Chemical Substances Regulations subsequently amended. (HCS) - Classification under the Globally Harmonized System of classification and labelling of chemicals (GHS)	
Emergency Overview	Colour: Liquid Clear Taste: None Physical State: Gas Form: Gas under pressure Odour: Stenched - Will have a pungent garlic or skunk-like smell	
Main Hazards	-The hazards due to the handling of LPG stem mainly from its extreme flammability -Vaporised LPG gas is highly flammable and can form an explosive mixtures with air -The flammability limits in air are between 1.5% and 9.5% by volume -The vaporised liquid can act as a simple asphyxiant by diluting the concentration of oxygen in the air below the levels necessary to support life	
Adverse Health Effects	- LPG has some degree of anaesthetic action and is mildly irritating to the mucous	

	membranes after prolonged inhalation of
	high concentrations
Chemical	- LPG is virtually non-toxic and is a stable
Hazards	gas
Biological	- Contact with the liquid phase of LPG with
Hazards Vapour	the skin can result in frostbite
Inhalation	- LPG is non-toxic. Prolonged inhalation could have an anaesthetic effect. Since it
IIIIaiation	can displace oxygen in the air it could also
	act as a simple asphyxiant
Skin Contact	- Liquid : Could cause serious cold burns.
	- Gas : No Known effect
Ingestion	- Liquid - Could cause serious cold burns
GHS	- Flammable gas (Category 1)
Classification	- Gas under pressure (Liquefied gas)
GHS Pictogram	A A
Ono i ictogram	
	W
	V V
CHC Clausel	Danasa
GHS Signal	Danger
Words	
Words Hazard	- H280 : Contains gas under pressure; may
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Words Hazard Statements	- H280 : Contains gas under pressure; may explode if heated H220 : Extremely flammable gas
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3. COMPOSITION OF INGREDIENTS	
Chemical name	Propane (C ₃ H ₈)
CAS No	74-98-6
UN No	1978
ERG No	115
Hazard class	2.1
Hazchem Warning	2A Flammable Gas
Chemical name	Butane (C ₄ H ₁₀)
CAS No	106-97-8
UN No	1075
ERG No	115
Hazard class	2.1
Hazchem Warning	2A Flammable Gas



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Chemical name	Propylene (C ₃ H ₆)
CAS No	115-07-01
UN No	1977
ERG No	115
Hazard class	2.1
Hazchem Warning	2A Flammable Gas
Product name	LPG (C ₃ H _{8 +} C ₄ H _{10 +} C ₃ H ₆)
Product name CAS No	LPG (C ₃ H ₈₊ C ₄ H ₁₀₊ C ₃ H ₆) 68476-85-7
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CAS No	68476-85-7
CAS No UN No	68476-85-7 1075
CAS No UN No ERG No	68476-85-7 1075 115

4. FIRST AID		
Eye contact	- Rinse the eye with water immediately. Remove contact lenses, if present and easy to do. Continue rinsing. Flush thoroughly with water for at least 15 minutes. Get immediate medical assistance. If medical assistance is not immediately available, flush an additional 15 minutes and seek medical attention as soon as possible.	
Skin Contact	 Contact with evaporating liquid may cause frostbite or freezing of skin. In case of frostbite spray with water (DO NOT USE HOT WATER) for at least 15 minutes. Apply a sterile dressing. Seek medical attention. 	
Ingestion	- Not an expected route of exposure	
Inhalation	- In high concentrations may cause asphyxiation. Symptoms may include loss of mobility/consciousness. Victim may not be aware of asphyxiation. Remove victim to uncontaminated area wearing self contained breathing apparatus. Keep victim warm and rested. Seek medical advice. Apply artificial respiration if breathing stopped.	
Most important symptoms and effects, both acute and delayed:	 Respiratory arrest. Contact with liquefied gas can cause damage (frostbite) due to rapid evaporative cooling. Loss of co-ordination. In low concentrations may cause narcotic effects. Dizziness. Headache. Unconsciousness. Nausea, vomiting. 	

5. FIRE-FIGHTING	
Suitable extinguishing	- Water Spray or Fog Dry powder. Foam.
media	 NOTE: with dry powder it is essential to have complete coverage of the fire to prevent flash back.
Unsuitable extinguishing media:	- Carbon dioxide - Hand units are suitable for small fires only

Specific Hazards	 EXTREMELY FLAMMABLE GAS. May explode in a fire. Incomplete combustion may form carbon monoxide
Special fire fighting procedures:	- In case of fire: Stop leak if safe to do so. Do not extinguish flames at leak because possibility of uncontrolled explosive reignition exists. Continue water spray from protected position until container stays cool. Use extinguishants to contain the fire. Isolate the source of the fire or let it burn out.
Special protective equipment for firefighters:	- Firefighters must use standard protective equipment including flame retardant coat, helmet with face shield, gloves, rubber boots, and in enclosed spaces, self contained breathing apparatus.

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6. ACCIDENTAL RELEASE		
Personal precautions, protective equipment and emergency procedures:	 Evacuate area. Provide adequate ventilation. Consider the risk of potentially explosive atmospheres. In case of leakage, eliminate all ignition sources. Stop leak if safe to do so. Monitor the concentration of the released product. Prevent from persons entering sewers, basements and workpits, or any place where its accumulation can be dangerous. Wear self-contained breathing apparatus when entering area unless atmosphere is proved to be safe. 	
Environmental Precautions	- Prevent further leakage or spillage if safe to do so.	
Methods and material for containment and cleaning up:	-Provide adequate ventilation. Eliminate sources of ignition.	

7. HANDLING AND STORAGE

Safe Handling	-Only experienced and properly instructed persons should handle gases under pressure. Use only properly specified equipment which is suitable for this product, its supply pressure and temperature. Purge system with dry inert gas (e.g. helium or nitrogen) before gas is introduced and when system is placed out of service. Purge air from system before introducing gas. Containers, which contain or have contained flammable or explosive substances, must not be inerted with liquid carbon dioxide. Assess the risk of a potentially explosive atmosphere and the need for suitable equipment i.e. explosion-proof. Take precautionary measures against static



Conditions

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SAFETY DATA SHEET (SDS) LPG

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discharges. Keep away from ignition sources (including static discharges). Provide electrical earthing of equipment and electrical equipment usable in explosive atmospheres. Use non-sparking tools. Refer to supplier's handling instructions. The substance must be handled in accordance with good industrial hygiene and safety procedures. Ensure the complete system has been (or is regularly) checked for leaks before use. Protect containers from physical damage; do not drag, roll, slide or drop. Do not remove or deface labels provided by the supplier for the identification of the container contents. When moving containers, even for short distances, use appropriate equipment eg. trolley, hand truck, fork truck etc. Secure cylinders in an upright position at all times, close all valves when not in use. Provide adequate ventilation. Suck back of water into the container must be prevented. Do not allow backfeed into the container. Avoid suckback of water, acid and alkalis. Observe all regulations and local requirements regarding storage of containers. When using do not eat, drink or smoke. Store in accordance with local/regional/national/international regulations. Never use direct flame or electrical heating devices to raise the pressure of a container. Leave valve protection caps in place until the container has been secured against either a wall or bench or placed in a container stand and is ready for use. Damaged valves should be reported immediately to the supplier Close container valve after each use and when empty, even if still connected to equipment. Never attempt to repair or modify container valves or safety relief devices. Replace valve outlet caps or plugs and container caps where supplied as soon as container is disconnected from equipment. Keep container valve outlets clean and free from contaminates particularly oil and water. If user experiences any difficulty operating container valve discontinue use and contact supplier. Never attempt to transfer gases from one container to another. Container valve guards or caps should be in place. -All electrical equipment in the storage areas should be compatible with the risk of a potentially explosive atmosphere. -Segregate from oxidant gases and other oxidants being stored. -Containers should not be stored in conditions likely to encourage corrosion. -Stored containers should be periodically checked for general conditions and leakage.

-Container valve guards or caps should be in

-Store containers in location free from fire
risk and away from sources of heat and
ignition.
-Keep away from combustible material.
-Keep container below 50°C in a well
ventilated place.

B. EXPOSURE CONTROLS	
Occupational Exposure Hazards (HCS)	 None of the components have assigned exposure limits As vapourised LPG is a simple asphyxiant, avoid any areas where spillage has taken place. Only enter once testing has proved the atmosphere to be safe.
Engineering Control Measures	- Consider a work permit system e.g. for maintenance activities Ensure adequate air ventilation. Provide adequate general and local exhaust ventilation. Keep concentrations well below lower explosion limits. Gas detectors should be used when quantities of flammable gases or vapours may be released Systems under pressure should be regularly checked for leakages. Product to be handled in a closed system Use only permanent leak tight installations (e.g. welded pipes). Take precautionary measures against static discharges. Gas detectors should be used when toxic quantities may be released.
Personal Protection	 A risk assessment should be conducted and documented in each work area to assess the risks related to the use of the product and to select the PPE that matches the relevant risk. The following recommendations should be considered. Keep self contained breathing apparatus readily available for emergency use. Personal protective equipment for the body should be selected based on the task being performed and the risks involved.
Eyes	-Wear safety glasses with side shields or face shield.
Hands	 Protective gloves against mechanical risks. Wear gloves while handling containers Protective gloves against cold. Protective gloves should be used if there is a risk of direct contact or splash.
Body protection: Feet	- Wear fire resistant or flame retardant clothing.
1 361	-Wear safety shoes while handling containers

9. PHYSICAL AND CHEMICAL PROPERTIES Chemical Name LPG



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Chemical Symbol	C ₃ H _{8 +} C ₄ H _{10 +} C ₃ H ₆
Physical state	Gas
Form:	Liquefied gas
Colour:	Colourless
Odour:	Mercaptan like (Will have a pungent garlic or skunk-like smell)
Odour Threshold:	Odour threshold is subjective and is inadequate to warn of over-exposure.
pH:	Not applicable.
Melting Point:	-187.6 °C Experimental result, Key study
Boiling Point:	-42.1 °C to 0.5°C (1,013 hPa)
Sublimation Point: Not applicable.	
Flash Point: -104 °C	
Evaporation Rate: Not applicable.	
Flammability (solid, gas):	Flammable Gas
Flammability limit - upper (%):	9.5 %
Flammability limit - lower(%):	1.5 %
Vapour pressure:	1053 kPa (40 °C)
Vapour density (air=1)	1.5 to 2.0 (mixture dependent) (0 °C) AIR=1
Relative density:	0.536 (15 °C)
Solubility(ies)	
Solubility in Water:	Slightly soluble
Partition coefficient (n-octanol/water):	2.36
Autoignition Temperature:	450 °C to 540 °C (mixture dependent)
Decomposition Temperature:	Not available
Viscosity	Not available
Kinematic viscosity:	Not available.
Dynamic viscosity:	Not available
Explosive properties:	Not applicable
Oxidising Properties:	Not applicable
Molecular weight	44.09 g/mol(C ₃ H ₈) 58.1 g/mol(C ₄ H ₁₀)
Minimum ignition energy:	0.25 mJ

10. STABILITY AND REACTIVITY	
Reactivity	-No reactivity hazard other than the effects described in sub-sections below.
Chemical stability	-Stable under normal conditions.
Possibility of hazardous reactions	-Can form a potentially explosive atmosphere in air. May react violently with oxidants.

Conditions to avoid	 -Keep away from heat, hot surfaces, sparks, open flames and other ignition -sources. No smoking. - Never use cylinders as rollers or supports; or for any other purpose than storage.
Incompatible Materials	-Air, oxidisers. Chlorine dioxide For material compatibility see latest version of ISO-11114.
Hazardous Decomposition of Products	-Under normal conditions of storage and use, hazardous decomposition products should not be produced.

11.TOXICOLOGICAL INFORMATION	
Acute Toxicity	-Based on available data, the classification criteria are not met.
Skin & eye contact	- Based on available data, the classification criteria are not met.
Chronic Toxicity	 Based on available data, the classification criteria are not met.
Carcinogenicity	- Based on available data, the classification criteria are not met.
Mutagenicity	- Based on available data, the classification criteria are not met.
Reproductive Hazards	-Based on available data, the classification criteria are not met.

12.ECOLOGICAL INFORMATION	
Toxicity	- No known ecological damage caused by this product.
Persistence and degradability	The substance is biodegradable. Unlikely to persist
Bioaccumulative potential	 The subject product is expected to biodegrade and is not expected to persist for long periods in an aquatic environment.
Mobility in soil	 Because of its high volatility, the product is unlikely to cause ground or water pollution.
Ecology - soil	 Because of its high volatility, the product is unlikely to cause ground or water pollution.
Results of PBT and vPvB	- Not classified as persistent, bioaccumulating and toxic (PBT).
assessment	 Not classified as persistent, very persistent and very bioaccumulating (vPvB).
Other adverse effects	- May cause pH changes in aqueous ecological systems.
Effect on ozone layer	- None



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Effect on the global warming	- Globa
	- Conta discha

- Global warming potential: 3
- Contains greenhouse gas(es). When discharged in large quantities may contribute to the greenhouse effect.

13. DISPOSAL CONSIDERATIONS	
Disposal Methods	 Disposal of LPG, as with other gases, should be undertaken only by personnel familiar with the gas and the procedures for disposal. Contact the supplier for instructions. In general, should it become necessary to dispose of LPG, the best procedure, as for other flammable gases, is to burn them in any suitable burning unit available in the plant. This should be done in accordance with appropriate regulations
Disposal of Packaging	- The disposal of cylinders must only be handled by the gas supplier

14. TRANSPORT INFORMATION	
Road Transportation	
UN No.	1075
Shipping Name	Liquid Petroleum Gas (LPG)
ERG No.	115
Class	2.1
Subsidiary Risk	Flam gas 1
Hazchem Warning	2A-Flammable gas
Sea Transportation	
IMDG	1075
Shipping Name	Liquid Petroleum Gas (LPG)
ERG No.	115
Class	2.1
Subsidiary Risk	Flam gas 1
Label	Flammable gas
Air Transportation	
ICAO/IATA Code	1075
Class	2.1
Subsidiary risk	Flam gas 1
Packaging	- Cargo: 200
instructions	- Passenger: Forbidden
Maximum quantity	- Cargo: 150 kg
allowed	- Passenger: Forbidden

15. REGULATORY INFORMATION	
SANS 11014:2010 Edition 1	Safety data sheet for chemical products - Content and order of sections
SANS 10228:2012 Edition 6	The identification and classification of dangerous goods for transport by road and rail modes
SANS 10234:2019 Edition 2	Globally Harmonized System of classification and labelling of chemicals (GHS)

SUPPLEMENT TO SANS 10234 Edition 1	List of classification and labelling of chemicals in accordance with the Globally Harmonized System (GHS)
	NATIONAL LEGISLATION: OHSA 85 OF 1993

16. OTHER INFORMATION

- Ensure all national/local regulations are observed.
- Ensure users and relevant persons understand the asphyxiation hazard.
- Regularly check supplier's information sources for updated versions of SDS's.
- Ensure all national/local regulations are observed.
- Ensure operators understand the flammability hazard. Users of breathing apparatus must be trained.
- Before using this product in any new process or experiment, a through material compatibility and safety study should be carried out

Revision Date 30/03/2022 v01

Bibliography

Compressed Gas Association, Arlington, Virginia Handbook of Compressed Gases - 3rd Edition Matheson. Matheson Gas Data Book - 6th Edition SANS 10265 – The classification and Labelling of Dangerous Substances

SANS 10234 - List of classification and labelling of chemicals in accordance with the Globally Harmonized System (GHS)

SANS 11014-1 - Safety data sheet for chemical products: Content and order of sections

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